

CHAPTER 2

REVIEW OF LITERATURE

Review of literature

This chapter reviews the literature on reported studies performed on coconut milk, pineapple juice, and blended beverage, and also explores the effect of high-pressure homogenization and thermal pasteurization on the shelf life of juice. The role of nanocellulose synthesized from various sources in the stabilization of Pickering emulsion, and the bioaccessibility of curcumin encapsulated in emulsions are also reviewed.

2.1. Plant based milk and its health benefits

Functional beverages derived from plant sources are gaining popularity due to shifts in dietary habits, increased instances of endocrine disorders, lactose intolerance, allergies to cow milk, and skin issues, such as acne. Among the plant-based milk alternatives, five are frequently consumed: coconut milk, soy milk, almond milk, rice milk, and oat milk.

Plant-based milk substitutes vary greatly in terms of their nutrient density both across and within varieties. Their nutritional qualities are further influenced by the kind of raw material utilised, the techniques of processing, the addition of sugar and oil, and the fortification of vitamins and minerals. For those with milk protein allergies, plant-based protein sources can serve as substitutes (Walther et al., 2022). Compared to cow's milk, plant-based milks are often linked to reduced environmental effects (such as greenhouse gas emissions and water usage) (Ramsing et al., 2023).

2.2 Coconut milk and its use in different products

Coconut is economically important and used in many traditional foods especially in Asia as a source of dietary fat. There are generally two varieties of coconut: the tall and dwarf varieties. While whole milk contains more protein than coconut milk, which is equally creamy, coconut milk contains no lactose, so it poses no problem for those who are lactose intolerant. Those who are lactose intolerant may experience problems like cramps, gas accumulation, diarrhoea, and bloating when consuming cow's milk (Boateng et al., 2023).

Coconut milk adds flavour and taste to products such as ice cream and biscuits (Kwao et al., 2023). The coconut milk or coconut skim milk is a major component of many different recipes (e.g. curry), desserts, and products such as coconut syrup coconut jam/spread, beverages, , coconut tofu, bakery products, (e.g. coco soy milk and pina colada) and coconut cheese (Seow and Gwee, 1997; Patil and Benjakul, 2018).

Table 2.1: Proximate composition of coconut milk

Constituent	Range (%)
Moisture	50-54.1
Ash	1.0-1.5
Protein	2.8-4.4
Fat	32.2-40.0
Carbohydrate	5.5-8.3

Source: (Seow et al., 1997)

2.3. The effect of fat content on coconut milk stability

The effect of fat content in coconut milk on stability and digestibility has been reported in various research articles. One study evaluated the emulsifying properties, stability, and digestibility of coconut milk with different fat contents, showing that as the fat content increased, the droplet size and viscosity also increased. In both raw and processed coconut milks, separation of the oil-in-water emulsion is typically regarded as an unacceptable physical fault (Simuang et al., 2004). Chiewchan et al. (2006) reported that temperature, fat dispersion, and particle size all have a major impact on the stability of foods high in fat, including milk, yoghurt, and cheese. To maintain the stability of the coconut milk emulsion, homogenization and the addition of appropriate emulsifiers are necessary before heat treatment in order to reduce fat globule size. The type and quality of homogenization and emulsifier used has an impact on coconut milk's stability. Research has also shown that coconut milk with fat content below 10% can withstand environmental factors such as storage, lipid oxidation, and freeze thaw (Chen et al., 2024).

2.4. Digestibility of coconut fat and protein

A study found that monosodium glutamate can increase the digestibility of proteins in coconut milk while inhibiting the digestion and absorption of fat (Wu et al., 2024). Another study found that coconut milk, with a high fat content of 27.69%, is fast digestible and more resistant to oxidation compared to dairy milk (Beegum et al., 2022). The fat content in coconut milk plays a significant role in its stability and digestibility, with higher fat content leading to increased droplet size, viscosity, and faster digestion. However, lower fat content can also provide stability and resistance to environmental factors. Benaissa et al. (2019) reported that there is better gastrointestinal digestion of the proteins in coconut milk than in cow milk.

2.5. Effect of different emulsifiers on stability of coconut milk

The effect of different stabilizers on the stabilization of coconut milk has been investigated by researchers. Some of the common stabilizers used to improve the stability of coconut milk includes protein, sorbitol esters, ethoxy esters, and sucrose esters. These are substances that help to stabilize the emulsion by reducing the interfacial tension between the oil and water phases (Ariyaprakai et al., 2013). Heating coconut milk above 80 °C can lead to the denaturation of proteins, resulting in the complete destabilization of the emulsion (Raghavendra & Raghavarao, 2010). Coconut protein, an amphiphilic molecule, can be adsorbed, unfolded, and rearranged at the oil–water interface, stabilizing the oil-in-water emulsion (Chen et al., 2023). The coconut milk emulsion can be separated by adjusting the pH between 3 and 5.6, and an increase in pH from 6 to 10 can also result in higher destabilization of the emulsion (Raghavendra & Raghavarao, 2010). Adding starch to coconut milk can increase its viscosity and improve emulsification (Arlai and Tananuwong, 2021). The interaction between CG (coconut globulin) and coffee polyphenols can change the lipophilicity of CG and facilitate the formation of a dense and thick interfacial film at the oil–water interface, leading to improved emulsion stability (Chen et al., 2023). These stabilizers can be used to improve the stability of coconut milk, which is essential for its use in various food applications. However, the choice of stabilizer depends on factors such as taste, flavour, and the specific application of the coconut milk product.

2.6. Extraction of coconut oil using different methods

Coconut oil is an edible oil called, also known as copra oil (CO), is produced from the flesh or kernel of mature coconuts that are obtained from the coconut palm (*Cocos nucifera*). It has various applications due to its high saturated fat content; it is slow to oxidize and, thus, resistant to rancidification lasting up to six months at 24°C without spoiling. However, if CO is not properly extracted, the nutritional and medical properties might be destroyed because the extraction method employed has a direct effect on both the quantity and quality of the oil produced (Famurewa et al., 2021). The extraction of coconut oil using different methods has been extensively investigated. One study by Oseni et al. (2017) used six different techniques of coconut oil extraction, including enzymatic, chilling and thawing, centrifugation, natural-fermentation, and induced-fermentation processes. Ghani et al. (2018) also studied the effect of extraction methods on the quality and stability of coconut oil. In a comprehensive review by Ng et al. (2021), various extraction methods of coconut oil were summarized, including fermentation, refrigeration, centrifugation, and enzymatic separations. Fermentation

is a low-cost process that does not require expensive equipment or materials, but it is time-consuming and requires minimal heat to preserve nutrients. The method maintains the natural coconut flavour, but it requires a longer time to extract the oil (Ng et al., 2021). Another study by Agarwal and Bosco (2017) discussed the extraction processes of virgin coconut oil (VCO), including cold and hot extraction methods. Cold extraction involves destabilizing coconut milk without heating, while hot extraction involves pressing clean, ground, and fresh coconut to yield coconut milk followed by heating at high temperatures. The extraction of coconut oil can be done through two primary methods: wet and dry methods. The wet method is generally considered more effective than the dry method, with a higher level of effectiveness reaching up to 73% compared to the dry method's 70%. In wet method, coconut milk is chilled at $< 4^{\circ}\text{C}$ and agitated for 15 min with a rotator. After removing the top layer of cream from the aqueous layer, it is taken out and allowed to defrost in a water bath set at 50°C . Centrifugation is then done for 45 min at 6000 rpm in order to further isolate the VCO from the aqueous layer (Ghani et al., 2018). However, the dry method is more commonly used due to its lower cost and ease of operation. In dry extraction process, the coconut kernel is heated in an oven in controlled condition or sun-dried to remove moisture content. The dried kernel is pressed mechanically to obtain coconut oil (Ghani et al., 2018). The wet method requires more equipment and labour, making it less practical for large-scale commercial production. The quality of coconut oil extracted through the wet and dry methods can vary depending on the processing conditions and equipment used (Nurma et al., 2022). VCO extracted through the wet method, for instance, is considered a high-quality product due to its natural freshness and lack of additives. VCO is suitable for human consumption in its natural state without refining and is known for its health benefits (Agarwal and Bosco, 2017).

2.7. Nutritional benefits of coconut milk

Abdul Rahim et al. (2020) reported that coconut milk is higher in medium-chain triglycerides like lauric acid (46–54%), caprylic acid (5–10%), capric acid (5–8%) and has a lower risk of clogging arteries compared to cow's milk (Jadhav and Annapure, 2023). Additionally, coconut milk contains soluble and insoluble fibers, antioxidants, and micronutrients such as vitamin C, vitamin B, iron, and phosphorus, making it a healthy alternative to cow's milk (Agdeppa and Zamora, 2022). In addition to its anti-viral, anti-microbial, and anti-carcinogenic properties, coconut milk is linked to fitness benefits. It is also abundant in minerals and nutrients, including iron, calcium, potassium, magnesium, and zinc (Tulashie et al., 2022). The antioxidant property of coconut milk that was measured using 1,1-diphenyl-2-picrylhydrazyl (DPPH) radical scavenging test and the ferric reducing power (FRAP) assay, showed that coconut milk has

more antioxidant qualities than cow's milk. DPPH and 2,2-azino-bis (3-ethylbenzothiazoline-6-sulfonic acid) diammonium (ABTS) assays were also used to test the antioxidant properties of the coconut kernel methanolic extracts as a function of maturity. The antioxidant activities increased up to 190 days from the date of pollination and then decreased or remained unchanged (Karunasiri et al., 2020).

2.8. Biochemical and nutritional properties of pineapple juice

Pineapple juice promotes to healthy living because it is a excellent source of vitamins, phenols, organic acids, and carbohydrate. Its juice has 81.2–86.2% moisture, 13–19% total solids of which sucrose, glucose and fructose are the main sugars, 0.4% fibre, 0.89% crude protein, 0.50% ash, and 16.19% carbohydrate. The juice is a rich source of vitamin C with a content of 17.60 mg/100 ml (Kiin et al., 2016; Bamidele et al., 2017). It is also rich in vitamins A and B, besides containing several minerals such as calcium, phosphorus, and iron (Jiménez and Capron, 2018). It is a precious reservoir of polyphenolic compounds such as p-coumaric acid, caffeic acid, ferulic acid, sinapic acid, p-coumaroylquinic acid, feruloyl glucose, p-hydroxybenzoic acid, p-hydroxybenzaldehyde, and syringic acid, which have been revealed to have antioxidative, antimutagenic, and anticarcinogenic effects and have beneficial roles against cardiovascular diseases and cataracts (Wen and Wrolstad, 2002). It also contains bromelain enzyme that promotes digestion, and other components that contribute to various health benefits due to their anticancer, antidiabetic, and antihypertensive properties. Fresh pineapple juice contains the following important minerals: iron (0.2–0.31 mg/100 ml), manganese (0.3–0.99 mg/100 ml), phosphorus (3.1–8.0 mg/100 ml), magnesium (12–15.4 mg/100 ml), and potassium (124–130 mg/100 ml) (Khalid et al., 2016). Numerous advanced processing methods have been applied to reduce bacterial contamination, extend shelf life, and preserve antioxidants, vitamins, and minerals in pineapple juice that includes ultrafiltration, pasteurisation, freeze-drying, reverse osmosis, microwave treatment, supercritical carbon dioxide, ohmic heating, thermosonication, pulsed electric fields, high hydrostatic pressure, high-pressure homogenization, and UV irradiation (Leneveu et al., 2020).

2.9. Blended beverages

Beverages are consumed for their refreshing effect, and thirst quenching and nutritional properties. However, in recent times, functional beverages have gained consumer liking for their ability to modulate or prevent diseases. Plants from different sources have been used for making plant based beverages, including the following groups: i) cereals (rice, oat, corn, spelt, rye, quinoa, kamut); ii) legumes (soy, peanut, lupin, cowpea), iii) nuts (almond,

coconut, hazelnut, pistachio, walnut); iv) seeds (sesame, flax, sunflower, pumpkin, hemp); and v) pseudocereals (quinoa, teff, amaranth). As the sources used in these beverages vary, their nutritional composition also varies considerably (Munekata et al., 2020).

Consumption of blended vegetable and fruit juice boosts antioxidants, vitamins, and minerals level in the blood, lowers homocysteine levels (a risk factor for heart diseases), increases immunity, and reduces inflammatory indicators (Grau et al., 2023). Abedi et al. (2014) developed raspberry juice-milk, an acidic dairy drink having two main phases, milk phase (pH: 6.6–6.7) and raspberry juice phase (pH: 3.2±0.1). Due to the low pH in this beverage, the problem of milk protein sedimentation was encountered. The influence of pectin, carboxymethylcellulose (CMC) and kappa-carrageenan on the stability of the milk-raspberry juice drink and their synergistic effects were investigated. According to the results obtained, the most suitable samples were those comprising of a combination of pectin and CMC, exhibiting greater stability and viscosity as compared to samples consisting solely of pure CMC (Abedi et al., 2014). The milk–sour cherry juice mixture was stabilized using gum tragacanth and Persian gum as adsorbing and guar gum as nonadsorbing hydrocolloids along with inulin to enhance their stabilising properties. The combination of inulin and guar gum was able to better stabilise the mixture at certain ratios and concentrations. Guar gum played the main role in the stabilisation of the juice by increasing the viscosity and forming gel network (Teimouri et al., 2018). Food processors are more interested in natural fruit or vegetable juice–skim milk mixtures because fruits, vegetables, and milk are the main sources of various bioactive compounds, natural antioxidants, as well as macro- and micro-nutrients (Zulueta et al., 2007; Calligaris et al., 2004). The most common fruits used in ready to serve blended beverages include grapes, gooseberries, litchi, pineapple, orange, and mango. These fruits are valued for their nutritional content, refreshing quality, pleasant flavour, and medicinal properties. They are a source of vitamins, minerals, energy, and fiber, which can help reduce the risk of many diseases. They also provide antioxidants and antimicrobial properties, which can serve as a good appetizer. The development of functional beverages can boost their sensory properties and nourishing properties (Rathinasamy et al., 2021).

2.10. Effect of citric acid addition in low pH fruit juices

Fruit juices are suitable for consumption by persons of all ages. Many fruit juices are acidic (pH 2.5 - 3.8), which is harmful to most bacteria. Citric acid is traditionally added to juice to lower pH and inhibit browning. Citric acid (0.3%), potassium sorbate (0.025%), sodium benzoate (0.015%), and sugar (10%) were the preservatives that were utilized to extend the

shelf life of sugar cane juice. The shelf life was extended to 15 days at room temperature (26 ± 2 °C) and 35 days at 10 °C because of the treatment (Mishra et al., 2011). The number of microorganisms was discovered to be below the detection limit throughout this time (Mishra et al., 2011). Citric acid concentration affects the pH of fruit juices. As the concentration of citric acid increases, the pH decreases, as citric acid is a weak triprotic acid. This decrease in pH can impact the quality and properties of the fruit juice (Fidaleo and Ventriglia, 2022). The citric acid concentration impacts properties like pH, titratable acidity, and sensory characteristics of the final product (Tua et al., 2018). The low pH created by citric acid in fruit juices creates an environment that is less conducive to the growth of spoilage microorganisms, thereby prolonging the shelf life of the juice. Citric acid contributes to the stability of fruit juices by maintaining their quality attributes and sensory characteristics, thus ensuring a longer shelf life (Humayun et al., 2014). Researchers have observed that citric (1–20% w/v) and lactic (0.3–4.0% v/v) acids inactivate *Yersinia enterocolitica* at various temperatures (4°, 20°, 40°, 8 °C). The duration, temperature, and acid concentration of exposure affects the inactivation impact of lactic and citric acids (Virto et al., 2005). Citric acid naturally exists in fruits and vegetables. However, it is not the naturally occurring citric acid, but the manufactured citric acid that is used extensively as a food and beverage additive (Sweis and Cressey, 2018). Addition of citric acid or ascorbic acid to pasteurized sugarcane juice restricted the degradation of total soluble solids, total sugars and reducing sugars during storage. Potassium metabisulphite further reduced changes in these parameters, likely due to suppressing microbial activity (Chauhan et al., 2002). Orange, lemon, ginger juice and honey are rich in citric acid and have been used as natural preservatives in fruit drinks and other fruit products. However, quality fruit juice needs to be safe for consumption for 3–9 months, which can be achieved by combining natural and synthetic preservatives like citric acid (Tiencheu et al., 2021). Fresh sugarcane juice spoils within 3–4 h at room temperature and 8 h during refrigerated storage. Adding citric acid, ascorbic acid, or potassium metabisulphite as preservatives significantly slowed the decrease in pH and increase in acidity during storage of sugar cane juice, with the effect being higher at room temperature (Chauhan et al., 2002).

2.11. Thermal treatment of juice

The most popular technique for extending the shelf life of liquid foods is thermal treatment, which inactivates microorganisms and enzymes. However, heat alters the antioxidant properties of the food, causes unfavourable changes in its physicochemical properties, and irreversibly loses nutritional compounds. Many juice processors have searched for alternatives

to thermal pasteurisation in order to help preserve unstable nutrients (Zulueta et al., 2013). Mild pasteurization and minimal processing have received considerable attention due to the growing consumer awareness towards fresh-like, nutritive, and healthy products. The Food and Drug Administration (FDA) mandates at least a 5-log decrease of the microorganism of concern in order to accept processing conditions, as fruit juices have been linked to many outbreaks (Yildiz et al., 2019). Choo et al. (2022) reported the effects of thermal pasteurisation on microbiological, antioxidant, and nutritional qualities of noni juice. During the course of eight weeks of refrigeration (4 °C) storage, the level of microbiological counts in thermal pasteurized noni juice remained within a reasonable range, but other parameters like antioxidant and organic acid values in thermally pasteurised noni juice were lower than the fresh sample. He et al. (2016) reported that the effects of thermal treatment and high-pressure homogenization on the juices of apples, oranges, and grapes were favourable because they increased the phenolic and antioxidant content of the juices. These findings suggested that thermal treatment was superior to high pressure homogenization in terms of improving fruit juice functional qualities like enhanced antioxidants, vitamins, and minerals. Thermal treatment disrupts the cell walls; it promotes the release of bound phenolic compounds from fruit cells and increases their concentration in the juice. The high-pressure homogenization ruptures the fruit cellular structure, which initiates the oxidative degradation of phenolic compounds by cytoplasmic polyphenol oxidase (He et al., 2016).

2.12. High pressure homogenization of juice

A lot of research is being done on emerging technologies to provide food items with the fresh qualities that modern customers want. High pressure is one of the methods used primarily to inactivate microorganisms or enzymes that may modify or harm fresh food items. It can be utilised in static (high hydrostatic pressure, HHP) and dynamic (high-pressure homogenization, HPH) forms. Utilising these cutting-edge technologies has the benefit that some of them are also referred to as nonthermal technologies as the processing temperature never goes over 40 °C. As a result, research on the impact of high pressure on the sensory, microbiological, and enzymatic properties of food has been done. The heat-labile pectinesterase in orange and grapefruit juices was rendered inactive by high pressure (500–600 MPa). After processing orange juice for 10 min at pressures between 500 and 900 MPa, the hazy stability remained (Welti et al., 2009). Due to the comminution brought on by the intense fluid-mechanical stresses generated on the pressurised (up to 350 MPa) liquid when flowing at high velocity through a micrometric gap (homogenization valve), HPH can be profitably applied to reduce

the particles and emulsion droplets dispersed in a liquid to finer dimensions. The numerous studies on HPH microbial inactivation attest to the fact that the comminution action of HPH may also be used to break the cell membranes of microorganisms floating in liquid. In this regard, HPH can be regarded as a nonthermal technology because, as demonstrated by numerous studies, microbial inactivation during HPH treatments is only weakly attributable to thermal effects, even with the temperature rise brought on by frictional heating in the homogenization valve (roughly 0.2°C/MPa) (Maresca et al., 2011). A patent from two decades ago stated that juice homogenised at high pressure (100 MPa) has a longer shelf life and less microbial activity than juice homogenised at regular pressures (Clark et al., 1993).

2.13. Serum separation in blended beverage

Fruit juices undergo fast sedimentation due to their natural composition. Juices do not contain substances that can stabilize the movement of particles. Sedimentation of juices may be considered a technological problem for the food industry. If the fruit juices are not homogeneous, the properties of the juice, such as viscosity and thermal conduction, may be different for all materials, damaging their processing. Also, sedimentation of juices may be a problem regarding the sensory (visual) acceptance of the consumer. During storage, these particles aggregate to form sediments due to the action of gravity (Silva et al., 2019). The physicochemical properties of foods depend on interactions of food components including proteins and polysaccharides. For many processed food products, proteins and polysaccharides serve as fundamental ingredients to improve stability, rheological properties, and mouth feel. Products like milk and soymilk mixed with fruit juice can be described as acidified liquid protein systems with physical stability and viscosity almost like natural milk. They are usually composed of a dairy based acidic fraction (fermented dairy base) or a neutral fraction (milk, soymilk, etc.), which are further acidified with an acidic medium (fruit juice, fruit pulp, fruit juice concentrate, etc.) where it can also contain some flavourings, sweeteners, and stabilizers (Abbasi and Mohammadi, 2013). If the protein and polysaccharide carry similar net charge, electrostatic repulsive forces occur, resulting in thermodynamic incompatibility between biopolymers, further leading to separation in protein- and polysaccharide-rich phases. The stability of casein-hydroxypropyl cellulose complexes was enhanced with the increasing homogenization pressure, especially for the complex containing high molecular weight hydroxypropyl cellulose. The apparent particle size of complexes was reduced from ~200 nm to ~130 nm when using 300 MPa, when compared to the non-homogenized controls (Ye and Harte, 2014). The effect of guar gum as stabilizer on physicochemical and organoleptic

properties were determined at different concentrations in kiwi fruit-based ready to serve beverage incorporated with lemongrass. Study results revealed that increasing the guar gum concentration increases the total soluble solids, titratable acidity, and viscosity. Organoleptic properties viz. mouthfeel and flavour were also significantly affected (Bochare et al., 2020).

2.14. Bioaccessibility and stability of blended beverages

Due to an increased interest in blended beverage, researchers have studied the *in vitro* gastrointestinal digestion of fruit juice (pineapple, orange, and kiwi) blended with soya milk and reported that water-fruit juice beverages favoured the bioaccessibility of phenolic compounds and hydrophilic antioxidants. Islam et al. (2021) developed a blended beverage of whey and pineapple juice, and shelf life of the blended beverage was investigated for sensory attributes, phase separation, acidity, and probiotic cell viability. The results revealed that the probiotic whey-pineapple beverage could be preserved for 42–56 days at 4°C with satisfactory acceptability and quality. Fruits contain several useful substances including antioxidants. Biological antioxidants have been defined as “compounds that protect biological systems against the potentially harmful effects of processes or reactions that cause extensive oxidations. It must be noted that various processing methods applied to food materials may have significant effects on their antioxidant potential and phytochemical bioaccessibility. For instance, the bioaccessibility and bioavailability of phenolic compounds could be affected by interaction with other macromolecules present such as proteins, carbohydrates, and lipids. The bioaccessibility of total phenolic compounds ranged from 13.52–26.49% and of flavonoids between 24.25–67.00%. The pomegranate juice and the juice of black grapes and aloe vera kept 58.12 and 50.36% of their initial anthocyanins content, while for the other samples less than 1.10% was established. As a result, a maximum of 30% remaining antioxidant activity was measured for some of the samples, but for most this was less than 10% (Mihaylova et al., 2021). A study found that the presence of biopolymers, such as gum Arabic, can enhance the bioaccessibility of bioactive compounds in freeze-dried orange juice (García et al., 2023). Fermentation was found to increase the bioaccessibility of polyphenols and carotenoids in mango cultivar juices, which in turn enhanced their antioxidant activity (Cele et al., 2022). The effect of *in vitro* gastrointestinal digestion on the bioaccessibility of phenolic compounds from fruit juices was studied. The study found that the bioavailability of phenolic compounds can be affected by the digestive process, with some phenolic acids showing increased liberation in digested samples (Mihaylova et al., 2021). Researchers investigated the ability of iron from almond and coconut beverages to be taken up by Caco-2 cells, which are a model of the human intestinal epithelium. The study found that both almond and coconut beverages significantly

increased ferritin induction when iron sulfate was added, suggesting that these beverages could either contribute directly to iron supply or indirectly by enhancing iron absorption (Silva et al., 2022).

2.15. Bioactive properties and bioavailability of curcumin

Curcumin [1,7-bis(4-hydroxy-3-methoxyphenyl)-1, 6-heptadiene-3, 5-dione], a low molecular weight hydrophobic polyphenol derived from turmeric rhizomes (*Curcuma longa* Linn.), is used as a spice in many foods and as a colouring agent (Ma et al., 2019). Curcumin is the main bioactive ingredient in turmeric. *Curcuma longa* contains 3-5% curcuminoids with four primary forms, curcumin (77%), demethoxycurcumin (17%), bisdemethoxycurcumin (3%), and cyclocurcumin (Araiza et al., 2018). The hydroxyl groups of the benzene rings, double bonds in the alkene parts and the central β -diketone moiety in curcumin are responsible for many chemical/biochemical reactions such as free radical scavenging, antimicrobial and antioxidant activities (Rao et al., 2016). Curcumin is one of the well-studied nutraceuticals with health benefits and researchers are interested in developing curcumin-based functional foods. The studies related to absorption, distribution, metabolism, and excretion of curcumin have revealed that curcumin has low solubility in gastrointestinal fluids (Ngwabebhoh et al., 2018). The bioavailability of curcuminoids, unfortunately, is low, mainly due to their water insolubility and rapid metabolism. It has been shown that bisdemethoxycurcumin (BD-Cur) and demethoxycurcumin (D-Cur) have similar bioactivities as curcumin (Yu and Huang, 2012). Various nanocarriers have been developed to increase the oral bioavailability of curcumin (Ma et al., 2019). The in vivo pharmacokinetics revealed that curcumin-encapsulated nanoparticles demonstrate at least 9-fold enhance in oral bioavailability when contrasted with curcumin regulated with piperine (Rincon et al., 2020). Permeation experiments on Caco-2 cell monolayers suggested that digestion-diffusion was the major absorption mechanism of curcumin within nanoemulsion (Yu and Huang, 2012).

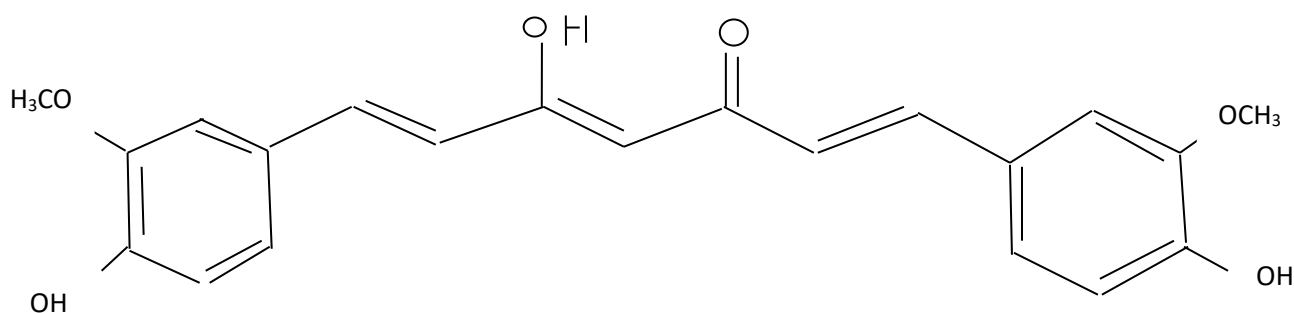


Fig 2.1: Structure of curcumin (Zhai et al., 2020).

2.16. Solubility of curcumin

One of the key areas of interest when it comes to curcumin dissolution is solubility enhancement. There are two primary approaches to consider: using substances to improve curcumin dispersibility and reducing the size of the particles to increase the surface area. The popular and effective strategies among them include inclusion complex creation, hydrophobic carriers, and surfactants (Tabanelli et al., 2021). Previous researchers have indicated that curcumin's low solubility and wettability result in poor dissolution, which in turn leads to poor bioavailability. A drug's solubility and/or dissolving rate determine its oral bioavailability, and dissolution may be a rate-determining step for the onset of a medicinal effect. For this reason, it is sometimes necessary to make attempts to improve the dissolution of drugs with poor water solubility (Modasiya and Patel, 2012). Luo et al. (2022) reported about the possibility of increasing curcumin absorption by excipient emulsions. Three distinct types of oils were used to create oil-in-water excipient emulsions: medium chain triglycerides (MCT), maize oil, and olive oil. Excipient emulsions can decrease cellular metabolism and increase trans-enterocyte absorption of curcumin, hence boosting its bioavailability. O/W emulsions were made with fish, flaxseed, coconut, sunflower, or maize oils. The solubility of curcumin increased with temperature at 100 °C (190–200 µg/ml) than at 30 °C (30–36 µg/ml) (Zou et al. 2016). Joung et al. (2016) reported that the solubility of curcumin was much higher in MCT oil (45% capric acid and 55% caprylic acid) as compared to coconut oil and olive oil. The solubility of curcumin in coconut oil, olive oil, and MCT oil was 0.1 mg/ml, 0.08 mg/ml and 0.25mg/ml, respectively.

2.17. Antimicrobial activity of curcumin

The traditional medical practices of Pakistan, Bangladesh, and Afghanistan, as well as the Ayurvedic and Unani systems, and traditional Chinese medicine, have all made extensive use of this rhizome. In the past, people have used turmeric as an antiseptic, antibacterial, anti-inflammatory, choleric, and carminative agent to treat burns and wounds, liver and gastrointestinal issues, respiratory conditions (such as sinusitis, asthma, coughing, and runny nose), anorexia, and rheumatism (Adamczak et al., 2020). Research has shown that curcumin has strain-specific activity, with a stronger effect against Gram-positive bacteria compared to Gram-negative bacteria. Studies have demonstrated that curcumin inhibits the growth of both Gram-positive and Gram-negative bacteria, with minimum inhibitory concentrations ranging from 31.25 to 5000 µg/ml. Notably curcumin has been found to be more effective against Gram-positive bacteria like *Staphylococcus aureus* and *Staphylococcus epidermidis*

(Adamczak et al., 2020). When combined with antibacterial drugs, curcumin has shown a synergistic relationship, enhancing the antibacterial effect against certain pathogens. Additionally, curcumin-based photodynamic treatment has been explored for its antibacterial efficacy, showing promising results in inactivating *Staphylococcus aureus* in juices, which can potentially be utilized to extend the shelf life of beverages. These findings highlight the potential of curcumin as a natural antimicrobial agent with diverse applications in combating bacterial infections and foodborne pathogens (Dai et al., 2022; Górski et al., 2022; Hussain et al., 2022).

2.18. Curcumin nanoemulsion

Curcumin nanoemulsions were developed with different surfactant concentrations followed by high pressure homogenization and finally incorporated in a milk system (Joung et al., 2016). Two immiscible liquids (oil and water) combined with emulsifiers, thickeners, jelly agents, and a host of additional food-grade substances stabilised the mixture to generate emulsions and nanoemulsions. Their sizes varied between 10 and 100 nm for nanoemulsions and between 100 nm and 100 μ m for emulsions, therefore their primary differences were in size (Tabanelli et al., 2021). Curcumin nanoemulsion-fortified milk showed lower lipid oxidation than unfortified milk. Curcumin nanoemulsion dissolved in coconut oil, sunflower oil or linseed oil were fed to weaning rats for 60 days (Sugasini and Lokesh, 2017). The rat fed with curcumin in linseed oil showed high level of curcumin in serum, heart, liver and brain. Solubility of curcumin in oil is low and it is also known that much larger amount of curcumin is dissolved in heated oil that remains soluble for long period of time. Therefore, Yu and Huang (2012) studied the metastable (120 h) solubility of curcumin in four different oils. About 1 mg/ml curcumin was dissolved in MCT oil, canola, coconut, and corn oil and percentage bioaccessibility of curcuminoids was determined after in vitro lipolysis in both fed and fasted states. The results showed that the metastable solubility and lipolysis of MCT oil generated the highest percentage of bioaccessibility ($72.1\pm 6.4\%$ and $72.2\pm 10.1\%$, respectively) followed by coconut oil and then canola and corn oils. Curcumin release was 10% from nanoemulsion developed using ultrasound during gastric digestion and 23% under intestinal conditions, indicating its slow release and increased bioavailability. Curcumin stability in traditional emulsion was shown to be somewhat higher after comparing curcumin oil-in-water nanoemulsion and emulsion; however, nanoemulsions demonstrated superior stability (Tabanelli et al., 2021). The pure curcumin is very susceptible to chemical degradation in alkaline aqueous solutions ($\text{pH} \geq 7.0$) and tends to crystallise out of aqueous acidic solutions.

These effects were attributed to modifications in curcumin's molecular structure at various pH values. The resulting curcumin crystals were comparatively big (10–50 μm). It caused them to sediment quickly. Curcumin increases the water dispersibility and chemical stability of O/W emulsions. After a month of incubation at 37 °C, emulsions maintained in an acidic environment (pH <7) were able to keep >85% of the curcumin, while emulsions placed in an alkaline environment having pH 7.0, 7.4, and 8.0 were able to retain 62, 60, and 53% of the curcumin, respectively (Kharat et al., 2017).

2.19. Processing conditions for nanoemulsions

Emulsions are created when a shearing mechanical force splits the dispersed phase into tiny droplets, exposing two immiscible phases to one another. Two types of sources of shear force are used to create emulsions: high energy sources use a device to apply the shear force, and low energy sources understand how the system's thermal or chemical energy changes to produce the desired surfactants and each liquid phase's physical properties (emulsion formation). A two-step procedure is needed to create nanoemulsions: first, coarse emulsions must be generated, and then the large droplets must be broken down into nano-size using high-pressure homogenization or ultrasonication (Wilson et al., 2022). Ultrasonication involves the use of ultrasonic waves to disrupt droplets into smaller sizes, while high-pressure homogenization utilizes mechanical devices like high-pressure valve homogenizers to create nanoemulsions. These methods are crucial for achieving small droplet sizes, enhancing stability, and improving functional properties of nanoemulsions. Additionally, the preparation of nanoemulsions can be categorized into low-energy and high-energy methods, with each method offering distinct advantages in terms of efficiency and droplet size control (Ashaolu, 2021; Aswathanarayan and Vittal, 2019; Çınar, 2017). Microfluidization is another method used to manufacture nanoemulsion. This instrument operates using a high-pressure displacement pump (500–20,000 psi) that forces materials through a chamber with many microchannels. The materials flow through the microchannels to an impingement location where they are transformed into minuscule particles. Other methods used are emulsion inversion point (EIP) method and phase inversion temperature (PIT) method. The EIP approach involves spontaneous emulsification using low energy, whereas PIT involves raising the emulsion system's temperature to alter the surfactant's solubilization pattern from hydrophilic to lipophilic. This results in the formation of bicontinuous microemulsions, which are then followed by emulsion inversion (Sutradhar and Amin, 2013).

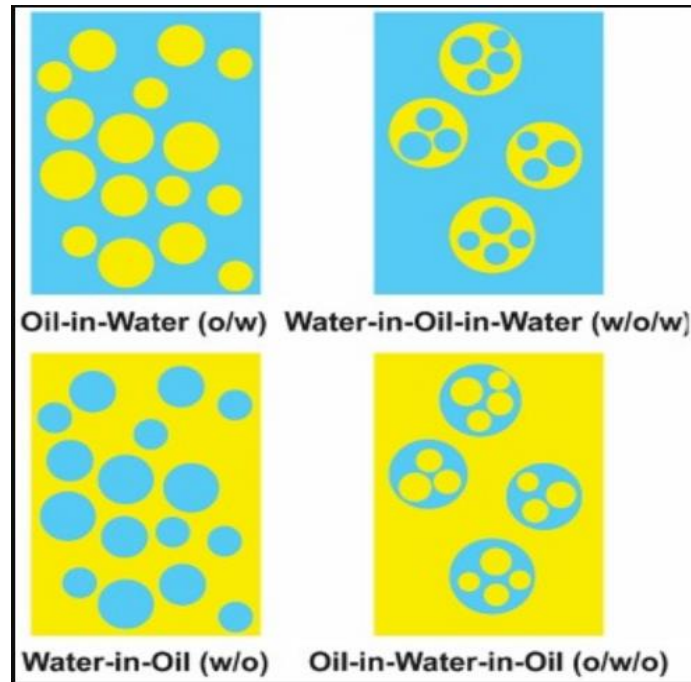


Fig 2.2: Different types of emulsion that are categorised by their continuous and dispersed phases (Wilson et al., 2022).

2.20. Nanoemulsion destabilizing mechanisms

When the oil and water phases of a nanoemulsion have different densities, the droplets in the nanoemulsion migrate as a result of gravitational forces, leading to gravitational separation. Depending on the relative densities of the phases, this can cause either sedimentation (movement below) or creaming (movement upward). There are several parameters that affect the rate of gravitational separation, including the viscosity of the continuous phase, droplet size, and density differential. Enhancing the viscosity of the aqueous phase and decreasing the particle size and density contrast can contribute to improving the stability of nanoemulsions (McClements et al., 2021).

The process by which two or more droplets combine to create clusters is known as flocculation. Repulsion forces like steric and electrostatic contacts may be outweighed by attractive forces like van der Waals, hydrophobic, depletion, and bridging attraction, causing this to happen. Larger droplets may develop as a result of flocculation, which can also occur before coalescence. While flocculation in concentrated nanoemulsions can hinder gravitational separation and cause the production of a semi-solid texture owing to Ostwald ripening, it can also speed up creaming or sedimentation in weak nanoemulsions (Liu et al., 2019). Depending on the nature of the forces, droplet aggregation can occur via coalescence or flocculation. Smaller droplets combine to form larger ones in the process of coalescence. Flocculation occurs

as two or more droplets cluster together. Flocculation facilitates coalescence by allowing droplets to remain in close proximity for extended durations. Due to droplet aggregation particle size increases, and it typically speeds up creaming or sedimentation. Oiling off, or the creation of an independent layer of oil on top of the nanoemulsion, can happen as a result of the attraction interactions between droplets. Several variables, including the type of forces at play and the makeup of the emulsifier being utilised, might affect coalescence (McClements et al., 2021). The process "oiling-off" occurs when individual layers of oil separate from the aqueous phase of a nanoemulsion. This is frequently observed in diluted nanoemulsions and might be caused by coalescence. By employing emulsifiers that improve stability and raising the aqueous phase viscosity, oiling-off can be minimised coalescence (McClements et al., 2021). Smaller droplets dissolve during the Ostwald ripening process, and the dispersed oil molecules spread to bigger droplets, enlarging them. This process, which is impacted by variables including the interfacial tension and oil molecule concentration, can result in a net increase in droplet size. Using an oil phase with limited water solubility or adding ripening inhibitors can prevent Ostwald ripening (Liu et al., 2019). In mixed colloidal dispersions with small and big droplets, the rate of Ostwald ripening may also be accelerated. Ostwald ripening is primarily caused by differences in particle sizes, hence emulsions with varying droplet sizes may experience an acceleration of this process. Furthermore, by reducing the diffusion path, direct contact between flocculated oil droplets may accelerate Ostwald ripening in emulsions (Salvia and McClements 2016).

2.21. Coconut oil nanoemulsion

Coconut oil is used in nanoemulsions due to its advantages over conventional emulsions. Nanoemulsions have a high bioavailability, enhanced long-term stability, and high optical clarity. The small droplet sizes in the range of 20-200 nm allow for better dispersion and improved solubility of the oil phase (Hasan et al., 2015; Wuttikul and Sainakham, 2022). Piriyaarasarth et al. (2012) made nanoemulsion formulations via conventional homogenization (13,500 rpm, 10 min) method using various surfactants and oil phase that consisted of 20–40% weight/weight concentrations of coconut oil, sunflower oil, and castor oil. The formulations containing coconut oil in the range of 20-40% w/w and the ratio of polysorbate 80 to sorbitan monooleate of 2:1 and 3:1 provided stable nanosized emulsions (100-500 nm). The results indicated that a decrease in the concentration of surfactant and an increase in the concentration of oil affected the stability of nanoemulsions (Piriyaarasarth et al., 2012). The nanoemulsions containing extract of mangosteen (*Garcinia mangostana* L.) rind were prepared using both

high-energy and low-energy emulsification methods, with VCO as the oil phase, Tween 80 and Span 80 as the oil-in-water surfactants, and distilled water as the aqueous phase. Results revealed that the VCO-based nanoemulsion was a promising carrier of mangosteen extract for topical formulation (Mulia et al., 2018). Coconut oil-based nanoemulsions have been studied for their potential in food applications, such as improving the compatibility, solubility, stability, and physicochemical equilibrium of various ingredients. For example, a coconut oil-based resveratrol nanoemulsion was optimized for use in food products. The optimization process involved using response surface methodology to study the effect of formulation and ultrasonication process factors on the globule size, polydispersity index (PDI), and zeta potential of the nanoemulsion (Kotta et al., 2021).

2.22. Effect of particle size on nanoemulsion stability

Ostwald ripening has an impact on micro and nanoemulsion stability. Adding a stabilizer, or a combination of them, can lower Ostwald ripening. It has been demonstrated that mixtures of costabilizers, both short and long chain, are effective in halting the deterioration of micro and nanoemulsions (Reyes et al., 2021). Food-grade Tween 80 and Span 80 were used as emulsifiers for the system based on the screening of previous experiments and the consideration of easiness. As the amount of Tween 80 increased, the particle size decreased. The effect of the difference in surfactant combination on particle size can be attributed to the change in HLB (hydrophilic-lipophilic balance of surfactant) values. HLB value indicates the affinity of the surfactant for the aqueous or oil phase and plays an important role in the phase transition of the low energy methods. Tween and Span are used as surfactants; while Tween has hydrophilic characteristics having HLB value >10 , Span has hydrophobic characteristics with HLB value <10 (Fang et al., 2022). The precise size of the oil droplets within an O/W emulsion depends on the type and concentration of emulsifier and oil used, as well as on the homogenization method employed. When an emulsion contains droplets with a mean diameter greater than 200 nm, it is usually categorized as a conventional emulsion (Choi and McClements, 2020). Different authors have given different maximum mean diameter of an emulsion; the highest limit has been suggested to be 100 nm (Dasgupta and Ranjan, 2018; Zhou et al., 2018), 200 nm (Mason et al., 2006; McClements and Rao, 2011), 500 nm (Gupta et al., 2016), or 1000 nm (Abramovits et al., 2010). This misconception is understandable given that crossing one of these particle size limitations does not significantly alter any of the physicochemical or functional characteristics of emulsions, such as optical qualities, creaming rate, rheology, or bioavailability. Rather, for various qualities, there usually is a rather slow shift that takes place

at different particle sizes. When the droplets are smaller than around 50 nm in diameter, a nanoemulsion may become nearly optically transparent; but, when the droplet size is less than 100 nm, the nanoemulsion becomes very stable against creaming. Products that must be optically clear, such as soft drinks and fortified waters, can employ this kind of nanoemulsion (McClements, 2011).

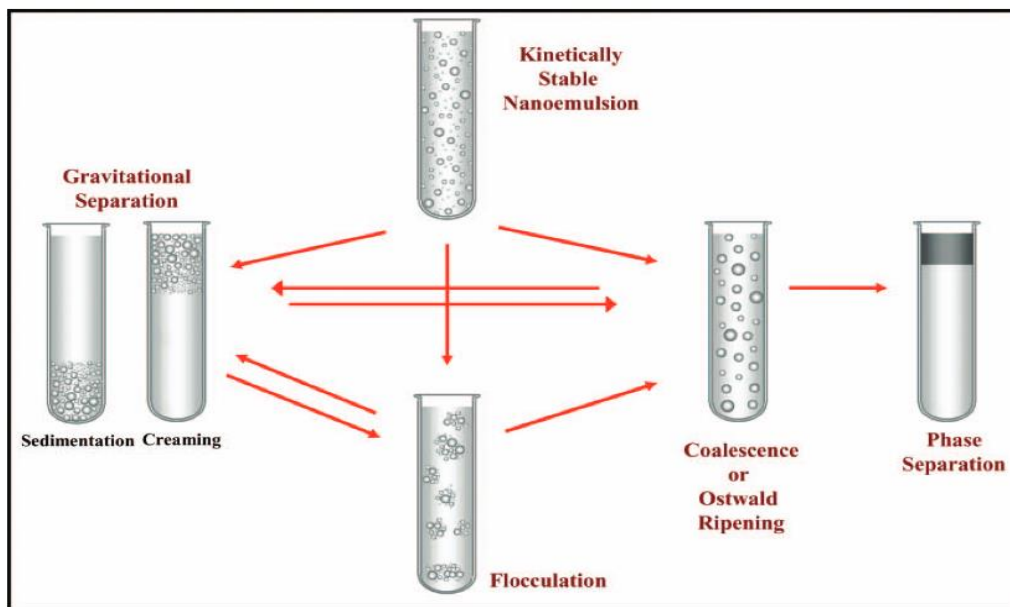


Fig 2.3: Nanoemulsions may breakdown numerous distinct physicochemical processes, including gravitational separation, flocculation, coalescence, Ostwald ripening, and oiling-off (McClements and Rao, 2011).

2.23. Pickering emulsions stabilized with various shapes of food grade particles

Food grade particles in various forms have drawn the attention of researchers in recent years (Zhang et al., 2021). The importance of the shape of particles in stabilizing Pickering emulsions, as well as the potential advantages of using non-spherical particles for this purpose were explored. Anisotropic particles with different geometrical shapes can generate stronger capillary forces and interfacial layers, leading to more stable emulsion systems (Li et al., 2022). Recent research has explored the use of various food-grade particles to stabilize Pickering emulsions, including polysaccharides, proteins, and inorganic particles. However, many of these particles do not naturally have the required dual wettability, so physical and chemical modification methods are needed. Particles can provide a space barrier between the two immiscible phases to prevent droplets' coalescence (Dickinson, 2012) and Ostwald ripening in Pickering emulsions. Compared to spherical particles, non-spherical particles like rods, discs, and fibers can potentially form more stable Pickering emulsions. This is because their

anisotropic shapes can generate stronger capillary forces at the interface (Li et al., 2022b). The particles with different shapes which have various particle densities, desorption energy, and capillary forces between adjacent particles largely affect the stabilization with different principles and application of emulsions. Particle shapes can be divided into regular spherical and irregular shapes with anisotropic morphology (Low et al., 2020). It was found that the principle of stable Pickering emulsion of rod-shaped particles was different from that of regular spherical particles.

2.24. Food applications of Pickering emulsions

Pickering emulsions have been the subject of academic interest in recent years due to their unique properties and potential applications in food systems. These emulsions are more physically and chemically stable than traditional emulsions, and they can offer a wide range of potential applications. They are made using a variety of techniques and can be stabilized by both organic and inorganic particles. While organic particles have a large molecular weight, inorganic particles are constrained by their indigestibility and lack of biodegradability. The stability of Pickering emulsions is influenced by factors such as wettability, particle shape, and size, and they can be used as stabilizers in food systems. The use of Pickering emulsions in food systems will improve the biological activity of food products (Potoroko et al., 2023). Pickering emulsions are more physically and chemically stable than traditional emulsions, which can be attributed to the solid particles that adsorb at the oil-water interface and form a dense film to prevent droplet aggregation (Chen et al., 2020; Potoroko et al., 2023). Pickering emulsions have a wide range of applications in food systems, including dairy and meat products, mayonnaises, and other food products (Potoroko et al., 2023). Pickering emulsions are more environmentally friendly than traditional emulsions, as they do not rely on synthetic emulsifiers (Klojdoová and Stathopoulos, 2022). Pickering emulsions can be made using biodegradable particles, which can be beneficial for food systems (Yang et al., 2017). Nanocellulose-stabilized Pickering emulsions have potential health benefits in food and beverage products due to their excellent stability and biocompatibility. These emulsions can be used in various food-related applications, such as delivery systems, food packaging materials, and fat substitutes. The unique properties of nanocellulose, including its nanosize, amphiphilicity, and biodegradability make it a suitable stabilizer for Pickering emulsions. The preparation processes are facile and easy to handle, as they are conducted in aqueous solutions where nanocelluloses are well dispersed and do not require any time-consuming solvent exchanging process (Fujisawa et al., 2017; Ji and Wang, 2023; Perrin et al., 2020). Pickering emulsions can be used for the encapsulation and delivery of biologically active substances,

which can improve the functionality of food products (Chen et al., 2020; Klojdová and Stathopoulos, 2022). However, there are technical challenges in using Pickering emulsions in food applications, such as the need for a proper choice of particle type and the requirement for particles to stabilize the emulsion (Linke and Drusch, 2018).

2.25. Plant-based emulsifier to reduce Tween level in emulsion

The research article by Phakthawat et al. (2023) discusses the use of hydrolysed rice glutelin as a plant-based emulsifier to reduce the need for synthetic emulsifiers like Tween in emulsion-based products, such as coconut milk. The study found that hydrolysed rice glutelin, a protein extracted from rice, can improve its emulsifying properties when partially unfolded by trypsin hydrolysis, leading to increased solubility and surface hydrophobicity. This improved emulsifying property can potentially replace Tween in emulsion-facing process stresses, making the emulsion more natural and environment friendly. Plant-based emulsifiers have gained significant attention in recent years due to their potential to enhance the functionality of food emulsions and contribution to sustainability. These emulsifiers, derived from natural sources such as plants, have been found to improve the stability, bioaccessibility, and oral bioavailability of encapsulated bioactive compounds in food emulsion. One study confirmed that plant-based ingredients, particularly saponins and lecithin, effectively produce stable O/W emulsions with flaxseed oil, offering opportunities for creating natural ingredient-based food emulsions (McClements et al., 2021). Another research article reviewed the use of natural emulsifiers in the design of emulsion-based delivery systems to protect bioactive compounds, discussing the properties and functionality of these systems (Teixé et al., 2023). In addition, plant-based emulsifiers have been found to be effective in improving the physicochemical properties of food emulsions, such as physical stability, microstructural characteristics, and rheological properties (Quezada et al., 2024). Soy lecithin, Quillaja saponin, vicilin and legumin, pea, lentil, faba bean proteins and methylcellulose films loaded with clove bud oil are some of the examples of plant-based emulsifiers for emulsion production (Deng, 2021; McClements et al., 2021; Quezada et al., 2024).

2.26. Nanocellulose

Nanocellulose is a natural and unique material extracted from native cellulose and has gained much attention due to its use as a biomedical material. Nanocelluloses are of three different types: cellulose nanocrystals (CNCs), cellulose nanofibrils (CNFs), and bacterial cellulose nanofibrils (BCNFs) (Benaissa et al., 2019). Nanocellulose, in the form of nanofibers and nanocrystals, can be obtained from various lignocellulosic waste materials like palm residues

and agave bagasse. This makes it a sustainable and eco-friendly option for food applications (Franco et al., 2023). Nanocellulose has a high surface area, aspect ratio, and water absorption ability, which contribute to its effectiveness as a stabilizing agent in food products. It can be used to modify the rheological behaviour of food systems (Perumal et al., 2022). CNCs and CNFs are both known to stabilize oil-water interface, forming the so-called Pickering emulsion which are surfactant free and highly stable (Jiménez and Capron, 2018). Pickering emulsion using medium chain triglycerides and Tween 80 and aminated nanocellulose particles with average particle size of ≤ 150 nm was obtained by Ngwabebhoh et al. (2018). CNCs, CNFs, and BCNFs have emerged as sustainable stabilizers/emulsifiers in food-related Pickering emulsions due to their favourable properties such as renewability, low toxicity, amphiphilicity, biocompatibility, and high aspect ratio (Ji and Wang, 2023). Some specific applications of nanocellulose in food stabilization include its use as an additive, food supplement, or in packaging materials. It can help improve the texture, stability, and shelf-life of various food products (Franco et al., 2023).

Since cellulose microfibrils are firmly attached in hemicellulose and lignin matrixes, it takes a lot of energy and chemicals to isolate pure cellulose. By hydrolysing rice straw holocellulose with sulfuric acid, amphiphilic holocellulose nanocrystals are produced. The yield of holocellulose (holoCNC) and α -cellulose isolated from rice straw using dewaxing-delignification method and alkaline treatment was 36.4 and 75.4%, respectively (Jiang and Hsieh, 2015). Both CNCs and holoCNCs could make O/W Pickering emulsions; however, the emulsion produced by the holoCNCs was 30% more emulsified and had emulsion droplets twice the size of those produced by the CNCs. HoloCNCs' distinct surface activity, amphiphilicity, and reduced self-assembling qualities provide desired traits that previously was only achievable through extra, frequently complex surface modification of CNCs (Jiang and Hsieh, 2015). Lei et al. (2022) reported that Pickering emulsion was successfully fabricated using bamboo shoot nanocellulose. With an average width of 56.37 nm and height of 7.44 nm, the nanocellulose that was isolated from bamboo shoots had significant promise as an emulsifier in the Pickering emulsions. Pomelo peels were used to create CNFs and CNCs using an easy method that involved sulfuric acid treatment and TEMPO oxidation. The longer fibrils in CNFs cause a gel structure to develop, because of which the stability of CNF-based Pickering emulsions was higher than that of emulsions stabilised with CNCs. The viscoelasticity of Pickering emulsions based on CNF was improved by higher oil fractions. Due to the bigger droplet size and higher viscoelasticity of the emulsion, the in vitro digestion data indicated that increased oil fractions lowered the degree of lipolysis.

Table 2.2: Extraction of nanocellulose using different samples and processing conditions

Nanocellulose samples	Type of nanocellulose	Processing conditions	References
Cotton linters	Nanocellulose crystals	acid hydrolysis	Morais et al. (2013)
Sisal fiber	Cellulose nanofibers	acid hydrolysis	Morán et al. (2008)
Husks of rice grains	Cellulose nanowhiskers	acid hydrolysis	Rashid and Dutta (2020)
Apple stem	Nanocellulose whiskers	acid hydrolysis	Phanthong et al. (2015)
Yerba mate sticks	Cellulose nanofibers	acid hydrolysis	Júnior et al. (2019)
Palm residues	Cellulose nanocrystal	acid hydrolysis	Mehanny et al. (2021)
Kenaf bast	Nanocrystalline cellulose	Microwave treatment and acid hydrolysis	Song et al. (2018)
Rice husk	Cellulose nanofibers and cellulose nanocrystals	High-pressure homogenization, acid hydrolysis and Combination of high-pressure homogenization along with acid hydrolysis	Samsalee et al. (2023)
Pineapple leaf fibers	Cellulose nanofiber	High-shear homogenization and ultrasonication	Mahardika et al. (2018)

Lycopene release exhibited a tendency like that of free fatty acid release, indicating that increased oil fractions were advantageous for regulating lycopene release during gastrointestinal digestion (Gao et al., 2023). After acid hydrolysis, HPH was employed as a post-treatment to modify the morphology and physical characteristics of the cellulose found in ginkgo seed shells. The resultant nanocellulose was then added to stabilise O/W Pickering emulsions. Small quantities of nanocellulose processed at 50 MPa was used to create emulsions with an oil phase ranging from 10 to 70% (v/v). Furthermore, the emulsions demonstrated benign stability under a variety of pH, ionic strength, and temperature conditions (Ni et al., 2020).

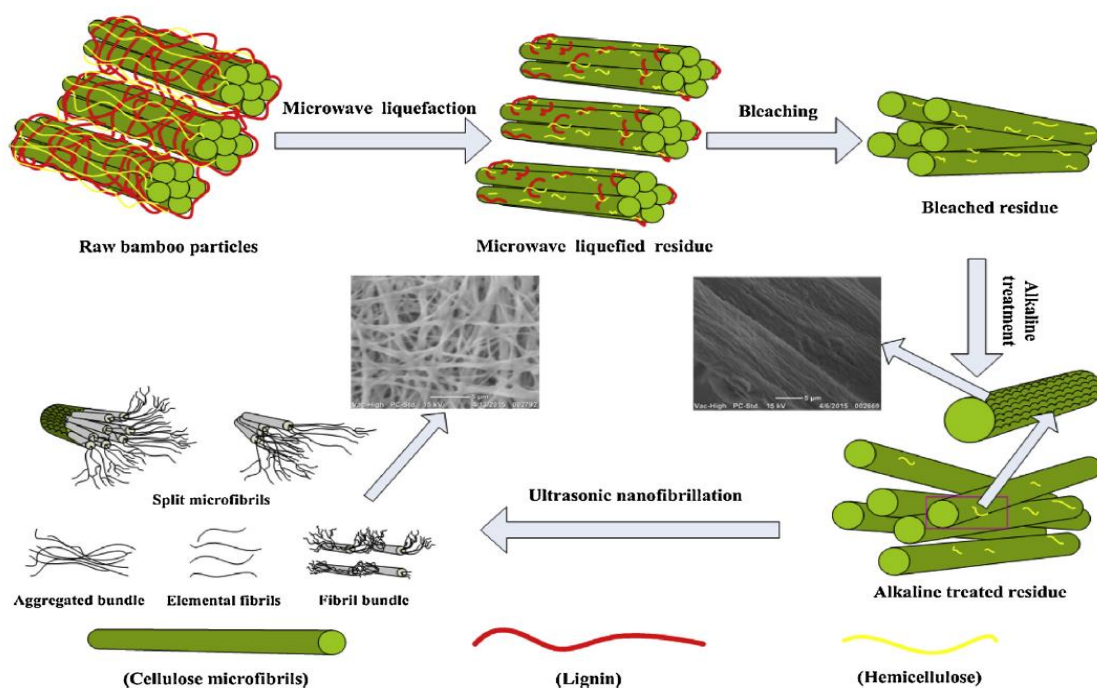


Fig 2.4: Graphical representation showing the formation mechanism of nanofibrils using combination of microwave and ultrasonication (Xie et al., 2016).

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